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# Physicochemical Characteristics of Fine Ambient Aerosol from Quasi-Rural Area in Southern Poland

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ABSTRACT: 24-hour samples of PM₁ (submicron particles, aerodynamic diameters ≤1 µm) and PM₁, (fine particles, aerodynamic diameters ≤ 2.5 μm) were collected at a quasi-rural area in Racibórz (1 January-30 June 2011). The samples were analysed for carbon (organic and elemental), water-soluble ions (Na<sup>+</sup>, NH, K, Mg<sup>2+</sup>, Ca<sup>2+</sup>, Cl, NO, SO, 20 and concentrations of 21 elements with a carbon analyser (Sunset Laboratory Ltd.), an ion chromatograph (Herisau Metrohm AG) and an Epsilon 5 spectrometer (PANalitycal), respectively. To perform the monthly mass closure calculations for PM<sub>1</sub> and PM<sub>2</sub>, chemical components were categorized into the organic matter (OM), elemental carbon (EC), secondary inorganic aerosol (SIA), crustal matter (CM), Na/K/Cl (sum of Na+, K+ and Cl- concentrations), other elements (OE) and unidentified matter (UM). The results show that the particulate matter (PM) concentrations and its chemical composition are mainly influenced by the anthropogenic emissions (coal, waste and biomass combustion in home furnaces and energy production based on hard and brown coal combustion). On the other hand, the secondary organic and inorganic (to a lesser extent) aerosols constitute a considerable part of the PM mass. The findings indicate serious problems related to the possible reduction in the fine PM concentrations in southern Poland. Even though places such as Racibórz can be formally qualified as rural sites, such areas in southern Poland do not necessarily have low PM concentrations (lower than in cities) or chemical compositions of fine particles different from those in cities.

Key words: PM, s, PM, rural background, Organic matter, Crustal matter, Toxic metals, Chemical mass closure

#### INTRODUCTION

It is a well-known fact that fine ambient particles play a key role in the global climate changes. Their negative influence on human health is particularly evident (Englert, 2004; Pope and Dockery, 2006; Seinfeld and Pandis, 2006; Paasonen et al., 2013). Obviously, their impact depends on their concentration levels in the air. The chemical composition of the particulate matter (PM) is another factor that directly influences the volatility, density, reactivity and toxicity of the atmospheric aerosol. It also has an indirect impact on the way the aerosol influences the environment, including human organisms. The knowledge of PM chemical compositions also facilitates the understanding of aspects such as time and space variations in ambient particle concentrations, sourcereceptor correlations and effectiveness of the emission reduction strategies. Thus, it is the basic information required for the identification of the PM emission sources.

The research into the PM chemical composition and the identification of its sources have been described in many studies over the last decade (such as Yin and Harrison, 2008; Canepari et al., 2009; Putaud et al., 2010; Spindler et al., 2010; Shahsavani et al., 2012; Choi et al., 2013; Crippa et al., 2013; Zhang et al., 2013). The PM, fraction is the least known when it comes to the chemical composition and sources. The PM concentrations and chemical compositions can vary significantly, which depends on the main emission sources and chemical reactions occurring in the atmosphere. For that reason, it is necessary to conduct the research into the PM concentrations and chemical compositions in as many sites as possible. It is particularly important for Central and Eastern Europe as the knowledge of the PM chemical compositions is very limited in these regions (Putaud et al., 2010). In Poland, the chemical compositions of PM<sub>2.5</sub> and PM<sub>10</sub> were investigated in the city of Zabrze, which represents an urban area of southern Poland (Rogula-Kozłowska et al., 2012). Moreover, the research into

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the concentrations of the PM, and PM, and their percentage in the total PM mass have also been conducted in Zabrze since 2005 (Klejnowski et al., 2007 and 2012). The research shows that the percentage of PM<sub>1</sub> and PM<sub>2.5</sub> in the total PM can be very high in the urban areas of southern Poland. For example, PM, made 52% of the PM<sub>10</sub> mass, whereas PM<sub>25</sub> constituted 68% of the PM<sub>10</sub> mass in Zabrze in June. For the same measurement point in December, the fine dust (PM<sub>2.5</sub>) made up 90% of the PM<sub>10</sub> mass, while the submicron dust (PM<sub>1</sub>) constituted 71% of the PM<sub>10</sub> mass (Klejnowski et al., 2012). Both this finding and the recognized chemical composition of PM<sub>25</sub> corroborate the fact that the fine dust in this area comes from anthropogenic sources. The studies also demonstrate that local sources, and particularly municipal emissions, are the main sources of fine dust in the Polish cities (Rogula-Kozłowska et al., 2012; Rogula-Kozłowska et al., 2013).

The aim of the following study was to understand concentrations and chemical compositions of PM, and PM<sub>2.5</sub> in the quasi-rural area in the south of Poland (Racibórz District, Commune of Racibórz). It also concerned the preliminary assessment of the PM, and PM<sub>25</sub> sources with the so-called chemical mass closure of ambient dust. The method compares sums of masses of the analysed dust components with the dust mass determined gravimetrically. Six classes of components were identified in this research, i.e. secondary inorganic aerosol (SIA), sum of sodium, chlorine and potassium (Na/K/Cl), organic matter (OM), elemental carbon (EC), crustal matter (CM) and other elements (OE). The difference between the sum of SIA, Na/K/Cl, OM, EC, CM, OE and the dust mass makes the so-called unidentified matter (UM). The classes were selected so that they could be compared with the data obtained at different European monitoring sites.

## **MATERIALS & METHODS**

The 24-hour PM<sub>1</sub> and PM<sub>2.5</sub> samples were collected with the Partisol 2020 dust sampler (Ruprecht & Patashnick) and PNS sampler (Atmoservice), respectively. The sampling took place between 1st January and 30th June 2011 in the Commune of Racibórz in southern Poland (Fig. 1). Racibórz is located in the Racibórz District in the south-west part of the Silesia Province. It is close to the border with the Czech Republic. Its geographical coordinates are 50°52 N and 18°142 E. The Racibórz District is an agricultural region. Moreover, agriculture is one of the main fields of its economic development as the agrarian land makes 66.30% of the area. A few plants (chemical, energy and agro-food industries) and trade centres are located only in the centre of Racibórz (approx. 6 km away from

the measurement point). The population density does not exceed 220 people per km² in the Racibórz District, whereas its level in the town of Racibórz is approx. 700 people per km². It is 3 to 5 times lower than in other cities of southern Poland. Most inhabitants live in the centre and its surrounding area. Low buildings with individual heating systems (biomass, coal, natural gas) dominate the landscape. The building development is compact in the centre and widely dispersed in the vicinity of the measurement point. The region of Racibórz is characterized by the following climatic data:

✓ Air temperature: average annual temperature −+8 °C; average temperature in January − -2.1 °C; average temperature in July −+18.0 °C.

✓ Number of freeze days: 100-110.

✓ Number of days with precipitation: 170 (including 45 days with snow).

✓ Winds along the Oder River (N-S): south winds dominate in autumn and winter, while north winds are more frequent in spring and summer; there is a large number of calm air periods.

 $PM_1$  and  $PM_{2.5}$  samples were collected on quartz fibre filters and Teflon membranes (Whatman Cat. No. 7592-104 and Whatman Cat. No. 1851-047, respectively) used in the alternate way every other day (the first filter was made of Teflon, then the quartz fibre one was used, and then the Teflon one was applied again, etc.). Before and after the exposition, all filters were conditioned in a weighing room (48 hours; relative air humidity  $45\pm5\%$ ; air temperature  $20\pm2$  °C) and weighed twice, with a 24-h period between the weighings, on a Mettler Toledo microbalance (resolution 2  $\mu$ g).

All 24-hour dust samples collected on the Teflon membranes underwent elemental composition analyses (Al, Si, Ti, V, Cr, Mn, Fe, Co, Ni, Cu, Zn, As, Se, Rb, Sr, Mo, Cd, Sb, Te, Ba, Pb). The elemental compositions of PM<sub>1</sub> and PM<sub>2.5</sub> were determined with energy dispersive X-ray fluorescence (EDXRF).

Two 1.5-cm<sup>2</sup> pieces were cut out from each of the quartz filters with dust. The first group underwent organic carbon (OC) and elemental carbon (EC) content analyses (Sunset Laboratories Inc.), whereas the other was examined for the contents of water-soluble ions (Cl-, NO<sub>3</sub>-, SO<sub>4</sub>-, Na+, NH<sub>4</sub>+, K+, Ca<sup>2+</sup>, Mg<sup>2+</sup>). To determine these ions, combined samples from all pieces collected within each month were prepared. The combined monthly samples (separate for PM, and PM<sub>2,5</sub>) were used to prepare water extracts. They were made through ultrasonizing the samples in 50 cm<sup>3</sup> of de-ionized water for 60 minutes at the temperature of 15°C. Then, they were shaken for approx. 12 hours (18°C, 60 r·min<sup>-1</sup>). The ion content in extracts was determined with a Metrohm ion chromatograph (Herisau Metrohm AG, Switzerland).



Fig. 1. Location of the sampling point

The parameters of the equipment (Sunset Laboratory carbon analyser, Epsilon 5 EDXRF spectrometer, Herisau Metrohm AG ion chromatograph), the detailed descriptions of the analytical procedures, and the results of the validation of the method are given in the studies (Rogula-Kozłowska *et al.*, 2012; 2013; Rogula-Kozłowska and Klejnowski 2013).

Analytically determined 24-hour masses of each component (from the 24-hour samples) were summed for each month, with the exception of ions as their combined monthly samples were analysed. The monthly masses of dust components were divided by the dust mass from which they were determined (sum of sample masses analysed in a given month). In this way, the average monthly percentage of each component in the dust was determined. The average monthly concentration of a component was its percentage in the dust multiplied by the average monthly dust concentration (arithmetic mean calculated from the 24-hour concentrations obtained in a given month). For the mass reconstruction (chemical mass closure), the PM<sub>1</sub> and PM<sub>2</sub> components were categorized into organic matter (OM), elemental carbon (EC), secondary inorganic aerosol (SIA), crustal matter (CM), Na/K/Cl, other elements (OE), and the rest—unidentified matter (UM). The mass [OM] of OM was assumed to be 1.8 [OC], where [OC]<sub>A</sub>—the analytically determined masses of OC (Grosjean and Friedlander 1975; Turpin and Lim 2001). The mass [EC] of EC was [EC] . SIA included PM, or

 $PM_{2.5}$ -bound  $SO_4^{2-}$ ,  $NO_3^-$ ,  $NH_4^+$ , so  $[SIA] = [SO_4^{2-}]_A + [NO_3^-]_A + [NH_4^+]_A$ .

The Na/K/Cl group included Na<sup>+</sup>, K<sup>+</sup> and Cl<sup>-</sup>, therefore  $[Na/K/Cl] = [Na^+]_A + [K^+]_A + [Cl^-]_A$ .

CM included  $CO_3^{2-}$ ,  $SiO_2$ ,  $Al_2O_3$ ,  $Mg^{2+}$ ,  $Ca^{2+}$ , FeO,  $Fe_2O_3$  and  $TiO_2$ ;  $[CM] = [CO_3^{2-}] + [SiO_2] + [Al_2O_3] + [Mg^{2+}]_A + [Ca^{2+}]_A + [FeO] + [Fe_2O_3] + [TiO_2]$ , where: [FeO] and  $[Fe_2O_3]$  were calculated stoichiometrically from  $[Fe]_A$  assuming a uniform Fe mass distribution between FeO and  $Fe_2O_3$  and  $[CO_3^{2-}] = 1.5[Ca^{2+}]_A + 2.5[Mg^{2+}]_A$  (Marcazzan *et al.*, 2001);  $[Al_2O_3]$ ,  $[SiO_2]$  and  $[TiO_2]$  were calculated stoichiometrically from the analytically determined  $[Al]_A$ ,  $[Si]_A$  and  $[Ti]_A$ ;  $[Ca^{2+}]_A$ ,  $[Mg^{2+}]_A$  were determined analytically.

OE included V, Cr, Mn, Co, Ni, Cu, Zn, As, Se, Rb, Sr, Mo, Cd, Sb, Ba, Pb and  $[OE] = [V]_A + [Cr]_A + [Mn]_A + [Co]_A + [Ni]_A + [Cu]_A + [Zn]_A + [As]_A + [Se]_A + [Rb]_A + [Sr]_A + [Mo]_A + [Cd]_A + [Sb]_A + [Ba]_A + [Pb]_A.$ 

UM was PM-OM-EC-SIA-Na/K/Cl-CM-OE, and [UM]=[PM]-[OM]-[EC]-[SIA]-[Na/K/Cl]-[CM]-[OE], where [PM]—determined gravimetrically mass of PM<sub>1</sub> or PM<sub>2</sub>,

The mass closure for  $PM_1$  and  $PM_{2.5}$  was examined separately for each month of the half-year research period in 2011.

#### **RESULTS & DISCUSSION**

In Racibórz, average monthly concentrations of PM<sub>1</sub> and PM<sub>2.5</sub> were 11.1-40.1 μg·m<sup>3</sup> and 12.0-75.0 μg·m<sup>3</sup>, respectively (Table 1). The average concentrations

of the analysed PM fractions obtained over the whole measurement period (arithmetic mean of average monthly concentrations) were 26.2 µg·m<sup>-3</sup> (PM<sub>1</sub>) and 37.9  $\mu$ g·m<sup>-3</sup> (PM<sub>2.5</sub>). The average monthly concentrations were at least twice as high in the cold season (January-March) than in the warm season (April-June). The maximum 24-hour concentrations reached the level of 124.1  $\mu$ g·m<sup>-3</sup> (PM<sub>1</sub>) and 152.3  $\mu$ g·m<sup>-3</sup> (PM<sub>25</sub>). The highest concentrations were observed in January and February. High PM concentrations are normally observed in the Silesia Province in this period. It is related to the occurrence of the lowest yearly temperatures and the impact of the emission from fossil fuel combustion (Pastuszka et al., 2010; Juda-Rezler et al., 2011). Higher 24-hour and average PM concentrations in the cold season (heating season) are observed in most Polish and European regions (Putaud et al., 2010; Spindler et al., 2010; Majewski et al., 2011; Zwoździak et al., 2012).

The results obtained in this study were compared with the results from other rural or quasi-rural sites in Europe (Table 2). Dissimilar geographical and meteorological conditions as well as types and activity of the emission sources have influence on the differences in the PM concentrations in the compared locations. Nonetheless, it is clearly visible that the PM<sub>1</sub> and PM<sub>2.5</sub> concentrations observed in Racibórz are significantly higher than those in other European regions. It is likely that specific emissions of ambient

dust and its precursors, characteristic particularly for southern Poland, are responsible for such a situation. The emissions originate from fossil fuel (mainly lowquality coal and fine coal), biomass, waste and rubbish combustion in home furnaces. The question of low emissions has already been partially solved in some Polish regions. The problem still exists in the south of Poland, particularly in the less well-off areas, such as the Racibórz District. Local inhabitants have virtually unlimited access to cheap, low-quality fuel that can be used in home furnaces. Importantly, the problem seems to have been gaining importance in recent years, in the era of the financial crisis. The location of Racibórz also influences the air quality. It lies in a close distance from the most highly urbanized and industrialized Polish region (Upper Silesia urban area) and highly industrialized Czech area (Moravia-Silesia).

Fine and submicron dust in Racibórz, similarly to other locations, is formed mainly from the carbonaceous matter (OC and EC) and secondary species ( $NO_3^-$ ,  $SO_4^{-2}$ ,  $NH_4^+$ ) – Tables 2 and 3.

The average PM<sub>1</sub>- and PM<sub>2.5</sub>-related concentrations of OC (obtained over the whole measurement period) were high and equalled to 8.46  $\mu$ g·m<sup>-3</sup> and 11.84  $\mu$ g·m<sup>-3</sup> (arithmetic mean from all monthly concentrations), respectively. The EC concentration was lower, but still relatively high in comparison with other chemical components. It was 1.89  $\mu$ g·m<sup>-3</sup> (PM<sub>1</sub>) and 2.72  $\mu$ g·m<sup>-3</sup> (PM<sub>2.5</sub>) for the whole

Table 1. Statistics on the 24-hour  $PM_1$  and  $PM_{25}$  concentrations in Racibórz, in particular months of the half-year research period in 2011

	January	February	March	April	May	June
$PM_1$						
Number of 24-hour concentrations <sup>a)</sup>	31	16	25	30	31	30
Minimum [μg·m <sup>-3</sup> ]	11.6	5.6	12.7	5.3	1.9	3.1
Minimum date	15-Jan	06-Feb	19-March	08-April	24-May	19-June
Maximum [μg·m <sup>-3</sup> ]	124.1	71.8	73.7	46.7	24.1	20.7
Maximum date	31-Jan	16-Feb	04-March	10-April	09-May	08-June
Arithmetic mean [μg·m <sup>-3</sup> ]	40.1	34.3	37.3	20.5	13.8	11.1
Standard deviation [μg·m <sup>-3</sup> ]	25.5	19.0	16.8	8.3	5.2	4.5
PM <sub>2.5</sub>						
Number of 24-hour concentrations a)	6	24	30	30	31	29
Minimum [μg·m <sup>-3</sup> ]	32.6	6.9	13.2	5.5	6.6	4.8
Minimum date	06- Jan	05-Feb	06-March	13-April	25-May	24-June
Maximum [μg·m <sup>-3</sup> ]	69.2	152.3	100.5	42.5	30.0	21.0
Maximum date	08- Jan	17-Feb	04-March	28-April	03-May	13-June
Arithmetic mean [μg·m <sup>-3</sup> ]	50.9	75.0	47.3	25.5	16.6	12.0
Standard deviation [μg·m <sup>-3</sup> ]	19.7	43.8	24.1	10.4	6.1	4.9

a) measurement time – at least 75% of the 24-hour period

measurement period. The cold-season concentrations of OC and EC were higher than the warm-season ones. It suggests that fossil fuel combustion was the main source of carbon compounds in Racibórz. The highest average monthly values of the OC/EC concentration ratio were observed in June. They ranged between 3.75 and 7.51 (PM<sub>1</sub>) and between 3.47 and 5.54 (PM<sub>2.5</sub>), which indicates the clear dominance of the organic carbon compounds over EC. This result can also suggest a significant influence of the secondary organic carbon (SOC) formation processes in the air on the concentrations and chemical composition of the fine dust in the examined area (Pio *et al.*, 2011).

SOC is formed as a result of the gaseous precursors reactions taking place in the atmosphere, which are described by over 500 different chemical reactions. The products of the gaseous precursors transformations condense on the inorganic salt particles and take part in the processes of cloud droplets formation (Grosjean and Friedlander, 1975; Seinfeld and Pandis, 2006). Indirect methods are most frequently used to estimate the SOC and primary organic carbon (POC) contents in the OC concentration (Castro, 1999).

The 24-hour  $PM_{2.5}$ -related SOC and POC concentrations ([SOC]<sub>24-hour</sub> and [POC]<sub>24-hour</sub>) in Racibórz (Fig. 2) were computed from the formula:

(Fig. 2) were computed from the formula:  

$$[SOC]_{24\text{-hour}} = [OC]_{24\text{-hour}} - [(OC/EC)_{min}]_{24\text{-hour}} \times [EC]_{24\text{-hour}};$$
 $[\mu g \cdot m^{-3}]$ 

$$[POC]_{24-\text{hour}} = [OC]_{24-\text{hour}} - [SOC]_{24-\text{hour}}; [\mu g \cdot m^{-3}]$$
where:

[OC]<sub>24-hour</sub> and [EC]<sub>24-hour</sub> are the 24-hour PM<sub>2.5</sub>-related OC and EC concentrations determined analytically; [(OC/EC)<sub>min</sub>]<sub>24-hour</sub> is the lowest 24-hour OC/EC ratio in the whole measurement period;  $[(OC/EC)_{min}]_{24-hour} = 3.58$ 

In Racibórz, the SOC percentage in the PM<sub>2.5</sub>-related OC was significant. It was slightly higher in the warm season when the insolation was stronger and the EC concentrations, absorbing the solar radiation, were lower. The high emission of organic gaseous precursors in the cold season influenced significantly the presence of the SOC fraction in the OC even though the conditions for the SOC formation were less favourable than in the warm season.

The following PM-related anions had the highest concentrations in the air:  $SO_4^{-2}$  (1.04-2.88  $\mu g \cdot m^{-3}$  for  $PM_1$  and 2.01-6.12  $\mu g \cdot m^{-3}$  for  $PM_{2.5}$ )  $NO_3^{-1}$  (0.37-2.92  $\mu g \cdot m^{-3}$  for  $PM_1$  and 0.61-5.54  $\mu g \cdot m^{-3}$  for  $PM_{2.5}$ ) and  $Cl^{-1}$  (0.01-1.81  $\mu g \cdot m^{-3}$  for  $PM_1$  and 0.05-2.56  $\mu g \cdot m^{-3}$  for  $PM_{2.5}$ ). On the other hand, for cations,  $NH_4^{-1}$  ambient concentrations were the highest. They were 0.17-1.16  $\mu g \cdot m^{-3}$  and 0.52-4.16  $\mu g \cdot m^{-3}$  for  $PM_1$  and  $PM_{2.5}$ .

respectively. Apart from Ca<sup>2+</sup> and Mg<sup>2+</sup>, the concentrations of water-soluble ions were higher in the cold season. They gradually decreased with each month passed. The most visible decrease in the average monthly concentrations concerned NO<sub>3</sub><sup>-</sup> (between January and June). The average monthly concentrations of Cl<sup>-</sup>, NO<sub>3</sub><sup>-</sup>, SO<sub>4</sub><sup>2-</sup>, Na<sup>+</sup>, NH<sub>4</sub><sup>+</sup> and K<sup>+</sup> were visibly correlated with the average monthly concentrations of EC (Table 4, linear correlation coefficient r, calculated at p<0.5). In conclusion, the above-mentioned ions probably originated from the combustion processes (particularly fossil fuel and biomass burning), similarly to EC (Terzi *et al.*, 2010; Đorđević *et al.*, 2012).

Significant correlations between the average monthly concentrations of virtually all ions (except for NO<sub>3</sub><sup>-</sup> and Na<sup>+</sup> as well as NO<sub>3</sub><sup>-</sup> and K<sup>+</sup>) were observed. Consequently, compounds such as KCl, NaCl, NH<sub>4</sub>Cl, NH<sub>4</sub>NO<sub>3</sub>, (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> and K<sub>2</sub>SO<sub>4</sub> were most likely the main compounds in which the given ions occurred.

 $\mathrm{NH_4NO_3}$  and  $(\mathrm{NH_4})_2\mathrm{SO_4}$  are the main inorganic products from transformations of gaseous precursors (SO<sub>2</sub>, NO<sub>x</sub> and NH<sub>3</sub>). In the air, SO<sub>2</sub> oxidizes to gaseous SO<sub>3</sub> or liquid H<sub>2</sub>SO<sub>4</sub>. Afterwards, it is neutralized to ammonium sulphate ((NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>) or ammonium bisulphate (NH<sub>4</sub>HSO<sub>4</sub>). NO<sub>x</sub> photochemically oxidises to HNO<sub>3</sub>. Then, it is neutralized to ammonium nitrate (NH<sub>4</sub>NO<sub>3</sub>). The ability to form these compounds and the speed of transformations depend on meteorological conditions and the proportion of the concentration of gaseous precursors of PM to the concentration of neutralizing compounds (Seinfeld and Pandis, 2006).

In Racibórz, the neutralization ratio NR (normal equivalent, ratio of NH<sub>4</sub><sup>+</sup> and SO<sub>4</sub><sup>2-</sup>+NO<sub>3</sub><sup>-</sup> concentrations) was generally lower for PM<sub>1</sub> than for PM<sub>25</sub>. Its values (calculated from average monthly concentrations) ranged from 0.34 (June, PM<sub>1</sub>) to 1.67 (January, PM<sub>25</sub>). The six-month means were 0.56 (PM<sub>1</sub>) and 1.26 (PM<sub>2.5</sub>). For PM<sub>1.5</sub> the amount of ambient NH<sub>2</sub><sup>+</sup> was not sufficient to fully neutralize ambient H<sub>2</sub>SO<sub>4</sub> and HNO<sub>3</sub> (NR<1). Consequently, it can be assumed that NH<sub>4</sub> did not form NH<sub>4</sub>NO<sub>3</sub> (at first, NH<sub>4</sub> neutralizes sulphates). It suggests the existence of PM, components other than (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> and NH<sub>4</sub>NO<sub>3</sub> in Racibórz, such as K<sub>2</sub>SO<sub>4</sub> in the cold season (biomass combustion). The low value of NR indicates that the aerosol pH could be acidic. In general, the PM acidity cannot be excluded at any month in Racibórz. It is a common assumption that if NH<sub>4</sub>+/SO<sub>4</sub>2-≥1.5 (molar ratio), then sulphuric acid is totally neutralized (Seinfeld and Pandis, 2006), but some authors report the PM<sub>2.5</sub> acidity even when NH<sub>4</sub>+/SO<sub>4</sub><sup>2</sup>->2 (Pathak et al., 2009; Huang et al., 2011).

Table 2. Mass concentrations of fine dust and percentages of selected components in the fine dust mass in the selected rural and quasi-rural areas in Europe

Ma Ma	DM moss	Canoto	Coo	Troop	o Culted Con Those Dismosted Oscanio	Organia				I Inidontified
Cita DM (cource of data)	concentration	Matter	Sea Falt	Hamente	Carbon	Oigaine Matter	$\mathrm{SO_4}^{2 ext{-}}$	$NO_3^-$	$NH_4^+$	Villuciumed
Site, Fivix(Source of data)	сопселитацоп [µg·m <sup>-3</sup> ]	(%)	3an (%)	(%)	(%)	(%)	(%)	(%)	(%)	(%)
Thüringer Wald (Germany), PM <sub>0.42-1.2</sub> (Gnauk et al., 2005)	9.6	-	-	-	5	61	22	91	8	38
Basel (Switzerland), PM <sub>2.5</sub> (Huegin et al., 2005)	18.9	9:9	1	3.6	8.5	23.7	21.8	16.6	10.8	3.8
Chaumont (Switzerland), PM <sub>2.5</sub> (Hueglin et al., 2005)	ĽĽ	8.8	ı	3.6	5.2	20.8	30.0	7.4	10.8	13.4
Hyytiälä (Finland), PM <sub>1</sub> (Saarikoski, 2008)	4.4	1	1	-	4	48		41 <sup>a)</sup>		7
20 km away Birmingham (UK); PM <sub>1</sub> /PM <sub>2,5</sub> (Yin and Harrison, 2008)	11.7/14.3	3.4/4.9	3.4/2.8	1	6.87.7	29.9/32.9	•	55.6/59.4		0.9/-7.7
Villar Arzobispo (Spain); PM <sub>2.5</sub> (Pey et al., 2009)	18.0	28.9	2.2	1	15.6	.6	13.9	7.2	6.7	25.5
Montseny (Spain); PM <sub>25</sub> (Pey et al., 2009)	13.6	9.6	1.5	ı	27.2	.2	20.1	8.8	8.1	24.7
Melpitz (Germany), PM <sub>1</sub> <sup>b) d)</sup> (Spindler et al., 2010)	10/11	•	ı	ı	6/01	6/01	20/18	£Z/5>	10/14	45/27
Melpitz (Germany), PM <sub>1</sub> <sup>c) d)</sup> (Spindler et al., 2010)	17/22	ı	ı	-	12.9	15/16	2 1/18	<2/14	6/6	41/27
Bologna (Italy), PM <sub>1</sub> <sup>d) e)</sup> (Carbone et al., 2010)	8.5/20.6	2,0 f)	2/0	1	48/43 g)	3 g)	27/8	8/34	13/14	1
Rome (Italy), $PM_1^{d e}$ (Carbone et al., 2010)	2.9/1.7	6/3 f)	5/2	ı	62/67 8)	7 g)	19/17	1/1	7/10	1
Rotterdam (Netherlands), PM <sub>2.5</sub> (Keuken et al., 2012)	15.8	4	4	4	11	20		33		24
Tourlos (Greece), PM <sub>1</sub> /PM <sub>2.5</sub> (Pateraki <i>et al.</i> , 2012)	14.2/19.2	1	1	1	2.7/2.3	20.2/17.4	29.0/27.0	2.0/4.0	8.1/6.3	ı
Salento (Italy), $PM_1/PM_{2.5}$ (Perrone et al., 2013)	15/25	0.9/0.8 <sup>h)</sup>	0.5/0.9	<0.5/<0.5	10/8.4	33.6/32.5	15.3/16.8	1.5/2.1	5/4.8	ı
Valverde del Camino (Spain), PM <sub>2.5</sub> (Milford et al., 2013)	21.56	20	4	ı	2 <sup>i)</sup>	22	12	8	3	22
Racibórz (Poland), PM <sub>1</sub> <sup>1)</sup> (this study)	37.2/15.1	1.6/4.1	5.6/3.7 <sup>k)</sup>	0.3/0.5	7.7/5.7	59.9/52.6	7.0/11.6	6.9/8.4	3.0/3.4	21.8/26.3
Racibórz (Poland), $PM_{2.5}^{-1}$ (this study)	57.7/18.0	1.8/6.1	6.4/3.6 <sup>k)</sup>	0.3/0.6	7.9/5.5	58.3/48.2	8.0/12.3	7.6/6.0	6.5/8.2	18.2/21.2

italics - data extracted from a chart

a) sum of water-soluble ions; b) air inflow from western sector; c) air inflow from eastern sector; d) summer/winter; c) samples taken during daytime; f) sum: (dust rich with Fe + Ca salts); \* sum of WSOM (water-soluble organic matter) and WINCM (water-insoluble carbonaceous matter); \* sum of Al and Fe; \* as black carbon (BC); \* ) cold

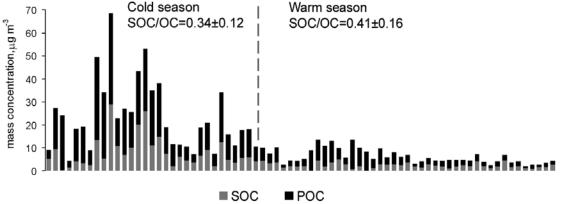
season/warm season; k) Na/K/CI

Table 3. Ambient concentrations of OC, EC, water-soluble ions [ $\mu g \cdot m^3$ ], and the remaining components [ $ng \cdot m^3$ ] bound to PM<sub>1</sub> and PM<sub>25</sub> in Racibórz (Poland)

	Jan	2011	Feb 2	2011	Marcl	h 2011	A pr il	2011	May	2011	June	2011
	PM <sub>2.5</sub>	PM <sub>1</sub>	PM <sub>2.5</sub>	$PM_1$	PM <sub>2.5</sub>	$PM_1$	PM <sub>2.5</sub>	PM <sub>1</sub>	PM <sub>2.5</sub>	$PM_1$	PM <sub>2.5</sub>	$PM_1$
OC	16.45	15.74	25.45	9.68	14.62	12.04	7.07	6.25	4.04	3.68	3.39	3.38
EC	4.74	3.47	4.59	2.58	3.92	2.61	1.55	1.38	0.88	0.86	0.63	0.45
Сľ	2.23	1.81	2.56	1.4	1.93	1.14	0.42	0.28	0.16	0.15	0.05	0.01
$NO_3$	2.83	2.09	5.54	2.92	4.76	2.68	1.78	2.69	0.86	0.77	0.61	0.37
SO 4 <sup>2-</sup>	4.47	2.88	6.12	2.73	3.34	2.22	2.59	2.41	2.03	1.81	2.01	1.04
Na <sup>+</sup>	1.14	0.51	0.45	0.34	0.39	0.26	0.36	0.35	0.28	0.27	0.28	0.18
$\mathrm{NH_4}^+$	4.16	1.14	3.62	1.07	3.50	1.16	2.6	0.94	1.33	0.43	0.52	0.17
K <sup>+</sup>	1.19	0.33	0.39	0.25	0.34	0.29	0.1	0.28	0.19	0.13	0.08	0.1
C a <sup>2+</sup>	0.09	0.06	0.38	0.04	80.0	0.07	0.09	0.02	0.03	0.07	0.06	0.04
$Mg^{2+}$	0.09	0.06	0.08	0.01	0.09	0.01	0.05	0	0.08	0.01	0.13	0.01
Al	50.37	35.7	42.65	30.91	56.67	28.58	29.88	23.77	41.45	32.45	66.26	34.06
Si	95.05	61.25	115.23	70.57	97.32	114.11	161.45	63	137.03	117.31	165.49	142.38
Ti	0.5	1.44	2.83	0.48	2.41	1.64	3.24	1.27	2.77	1.02	2.86	1.58
V	0.25	0.16	1.31	0.6	0.42	0.55	0.31	0.28	0.21	0.15	0.4	0.39
Cr	10.98	7.25	4.87	3.47	12.6	9.16	11.92	2.67	11.83	4.81	6.02	6.2
Mn	9.7	8.01	6.66	6.23	6.91	6.47	6.96	4.13	6.06	5.59	4.14	3.35
Fe	150.54	113.12	148.99	63.98	169.98	105.91	173.8	41.86	144.81	77.62	121.6	85.95
Со	0.91	0.28	0.1	0.1	0.15	0.13	0.17	0.14	0.34	0.24	0.01	0.01
Ni	7.44	6.47	8.09	3.47	20.52	13.3	21.8	2.25	18.22	3.32	8.73	4.92
Cu	8.47	7.39	7.37	5.03	5.60	4.96	4.56	2.86	5.01	4.61	6.65	5.02
Zn	56.3	59.19	105.91	54.04	65.74	53.82	49.82	19.92	34.05	28.38	19.42	18.9
As	6.97	6.16	13.12	5.75	6.71	5.53	5.89	2.3	3.85	3.7	2.44	2.15
Se	0.27	0.1	0.1	0	0.15	0.09	0.14	0.14	0.14	0.14	0.18	0.1
Rb	2.48	2.27	4.36	2.04	1.89	1.51	1.23	0.98	1.45	1.42	1.5	1.28
Sr	3.98	2.98	2.32	2.02	2.42	1.22	1.95	1.45	3.35	2.35	2.54	1.84
Мо	5.00	2.94	2.28	1.73	6.05	2.17	3.23	1.73	3.73	1.39	6.1	1.31
Cd	4.48	3.12	3.94	3.82	3 .54	3.24	2.82	1.83	2.42	1.67	2.36	1.9
Sb	5.25	4.45	9.48	4.19	4.57	2.01	3.94	3.03	2.94	2.15	1.67	1.19
Te	1.98	1.23	1.49	0.71	1.71	1.33	1.9	1.13	0.81	0.63	0.57	0.84
Ba	8.93	7.45	4.54	3.99	5.66	3.03	4.97	3.84	5.05	3.26	4.68	4.47
Pb	19.13	18.09	41.31	19.05	22.35	17.88	19.47	7.64	13.37	12.49	8.98	7.44
$(N H_4)_2 SO_4^{a)}$	6.2	4.2	8.4	3.9	4.6	4.3	3.6	3.4	2.8	1.6	1.9	0.6
NH <sub>4</sub> NO <sub>3</sub> b)	11.1	-	5.9	-	1 0.0	-	7.3	-	2.5	-	-	-

a) ambient concentration  $[(NH_4)_2SO_4]$  of  $(NH_4)_2SO_4$   $[\mu g \cdot m^{-3}]$  calculated as:  $[(NH_4)_2SO_4] = 1.38[SO_4^{-2}]_A$  if NR<1 and  $[(NH_4)_2SO_4] = 3.67[NH_4^{+}]_A$  if NR<1

b) ambient concentration [NH<sub>4</sub>NO<sub>3</sub>] of NH<sub>4</sub>NO<sub>3</sub> [μg·m<sup>-3</sup>] calculated as: [NH<sub>4</sub>NO<sub>3</sub>]=4.44[ex-NH<sub>4</sub>+]; [ex-NH<sub>4</sub>+]=[NH<sub>4</sub>+]<sub>A</sub> - 0.27[(NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>] if NR≥1



 $Fig.~2.~24-hour~concentrations~of~the~PM_{2.5}-related~SOC~and~POC~in~Racib\'orz~in~the~first~half~of~2011$ 

Table 4. Linear correlation matrix (p<0.5) for average monthly concentrations of water-soluble ions and EC bound to  $PM_1$  and  $PM_{2.5}$  in Racibórz (Poland)

	Сľ	NO <sub>3</sub>	$SO_4^{2}$	Na <sup>+</sup>	NH <sub>4</sub> <sup>+</sup>	$\mathbf{K}^{+}$	EC
Cl	1.00	0.84	0.86	0.63	0.77	0.66	0.99
$NO_3$	0.84	1.00	0.83	0.32	0.74	0.39	0.84
$SO_4^{2-}$	0.86	0.83	1.00	0.60	0.84	0.60	0.86
Na <sup>+</sup>	0.63	0.32	0.60	1.00	0.70	0.96	0.70
NH <sub>4</sub> <sup>+</sup> K <sup>+</sup>	0.77	0.74	0.84	0.70	1.00	0.69	0.82
K <sup>+</sup>	0.66	0.39	0.60	0.96	0.69	1.00	0.73
EC	0.99	0.84	0.86	0.70	0.82	0.73	1.00

For PM<sub>2,5</sub>, the NR was smaller than 1 only in June; in the remaining months, the NR was  $\geq 1$ . It shows the occurrence of NO<sub>3</sub>, SO<sub>4</sub> and NH<sub>4</sub> in the form of (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> and NH<sub>4</sub>NO<sub>3</sub>. For PM<sub>2.5</sub>, the NH<sub>4</sub>NO<sub>3</sub> concentrations were significantly higher than the (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> ones between January and April (Table 3). So, in Racibórz, ammonium nitrate occurred in the particles in the range of 1-2.5 µm. In the warm season, the weather conditions were not favourable to the formation of NH<sub>4</sub>NO<sub>3</sub> in the air. Even though a certain amount of NH, NO, might have evaporated, especially from the samples collected in summer, it is believed that the artefacts related to the semi-volatility of NH<sub>4</sub>NO<sub>3</sub> can be important only for the samples very rich in ammonium (Pathak et al., 2009). Generally, it may be expected that NO<sub>3</sub>- compounds with Ca<sup>2+</sup> or Na+ will occur in the summer period. For the whole measurement period (January-June) the linear dependence between average monthly concentrations of the PM<sub>1</sub>- and PM<sub>2</sub> -related NO<sub>3</sub> and Na<sup>+</sup> was weak (r=0.32, Table 4). For the warm season (April-June), however, the dependence was strong and characterized by the linear correlation coefficient r = 0.83.

The average concentrations of elements were highly diverse in Racibórz (Table 3). Generally, the highest concentration values were found for crustal elements (such as Al, Si and Fe). Low trace elements concentrations, bound to both PM<sub>1</sub> and PM<sub>25</sub>, were observed at the Racibórz measurement point. They were a few times lower than those measured in the cities of southern Poland in the previous years (Pastuszka et al., 2010; Rogula-Kozłowska et al., 2013). The Zn concentration was the highest (18.9-59.19 ng"m<sup>-3</sup> and 19.42-105.91 ng·m<sup>-3</sup> for PM<sub>1</sub> and PM<sub>2.5</sub>, respectively). The values of the Pb, Cr and Mn concentrations were lower (Table 3). Crustal elements were observed mainly in PM25, whereas PM1 was enriched with trace metals (such as Zn, Pb, Cr, Mn, Cu, Cd, Ni, As, Ba, V, Co) related to combustion and industrial processes (Querol et al., 2007; Rajšić et al., 2008). Obvious spatial variations were also observed in the ambient concentrations of elements. Most PM,-

and PM<sub>2.5</sub>-related elements had lower ambient concentrations in the warm period, except for crustal elements (such as Al, Si).

The total mass of the identified ambient dust components (OM, EC, SIA, Na/K/Cl, CM and OE) made 68–93% and 74–90% of the PM $_1$  and PM $_{2.5}$  masses, respectively (Fig.3). The percentage of the unidentified matter (UM) in the PM mass was generally higher in the warm season and practically did not depend on the particle size.

The lack of the complete chemical mass closure (reconstruction of the total mass determined gravimetrically) may be caused by many factors. Systematic errors committed during the sampling and gravimetric or chemical analyses can be one of the possible reasons for the discrepancy between the PM mass and the sum of its component masses. Moreover, it is possible that compounds that do not undergo chemical analysis are present in the ambient dust (Almeida et al., 2006). The lack of the complete mass closure is often assigned to the evaporation of the organic compounds and nitrates in the transport and PM sample storage processes (Rees et al., 2004). The observed differences can also result from uncertainties that arise from the assessment of the percentage of the chemical compounds that are not measured directly, particularly organic and crustal matter (Turpin and Lim, 2001; Rees et al., 2004; Almeida et al., 2006; Ho et al., 2006). The presence of water bound to PM particles is another important factor that influences the chemical balance of the ambient dust mass. Many studies prove that the water content in the total PM mass can be high (Tsyro, 2005; Canepari et al., 2012). For example, Tsyro (2005) observe that according to the model calculations the water percentage in PM (in the 50% RH conditions) is different in various European regions. It can range between 20 and 30% of the PM, mass. The percentage of water bound to the ambient aerosol depends on the content of hygroscopic compounds in the PM samples and is usually higher in the fine dust fractions (Ho et al., 2006).

OM was the most abundant compound, both in PM<sub>1</sub> and PM<sub>2,5</sub>; the OM percentage in PM was in the range of 48-71% and 44-61% for PM<sub>1</sub> and PM<sub>2,5</sub>, respectively. The OM content was significantly higher for both fractions in the cold season (particularly in January; Fig. 3). The higher OM content in the PM mass over the cold season resulted from high emissions of the organic carbon compounds, including the hydrocarbon emission from fuel combustion. The relatively high content of these components in the PM mass observed in the warm season results from the higher intensity of the SOC formation in the air in Racibórz (Fig. 2). It was also caused by the presence of the biological matter in the dust (Zhang *et al.*, 2009; Terzi *et al.*, 2010).

The EC percentage in the ambient dust mass was much lower (6 to 13 times) than the OM percentage. On average, it made approx. 7% of both PM<sub>1</sub> and PM<sub>2.5</sub> masses in the whole measurement period. Thus, the whole EC mass was accumulated in the submicron particles. Generally, the EC percentage in the dust mass and the EC concentrations in the air (Fig. 3) were higher in the cold season. It was related to the fact that there was an additional EC source in winter, namely fuel combustion in home furnaces. The highest EC percentage in PM (9%) was observed in January.

The secondary inorganic aerosol (SIA) made another important component of the fine PM in Racibórz. Its percentage was 2-11% and 5-16% for PM<sub>1</sub> and PM, masses, respectively. The highest SIA content was measured in May and June, when the meteorological conditions were favourable to the high intensity of transformations of aerosol gaseous precursors (Spindler et al., 2010). The secondary aerosol formation could be the possible explanation for increased concentrations of SO<sub>4</sub><sup>2</sup>, NO<sub>5</sub> and NH<sub>4</sub><sup>+</sup> in the warm period (Mkoma et al., 2009). On the other hand, fossil fuel combustion combined with the unfavourable meteorological conditions and the movement of air masses from more polluted areas contributed to high ambient concentrations of watersoluble ions in the cold season (Fig. 3, Table 3; Ho et al., 2006; Mkoma et al., 2009).

The percentage of the chloride, potassium and sodium (Na/K/Cl) masses in the fine dust mass was significant. The Na/K/Cl percentage was 2.4-7% and 3.5-9% for PM $_{\rm l}$  and PM $_{\rm l}$ , respectively (Fig. 3). The highest Na/K/Cl percentages in the PM $_{\rm l}$  and PM $_{\rm l}$ , masses were observed in January. Most of this mass was accumulated in the submicron particles.

As the sampling point in Racibórz was an inland site located far from the traffic sources, it was assumed that fuel combustion was the main source of the Na/K/Cl compounds. The above-mentioned analysis of the

correlations between Na<sup>+</sup>, K<sup>+</sup>, Cl<sup>-</sup> and EC corroborated this finding. As Racibórz is a quasi-rural location, the high percentage of the Na/K/Cl compounds in the fine dust mass could also be brought about by additional sources of the alkaline dust, such as grass burning and grass and plant waste combustion in the surrounding meadows, fields and gardens.

The CM percentage in the PM, and PM, depended largely on the season. It was low in the cold season and did not exceed 2%. The CM content started to rise gradually in April and reached its highest level in June (6% in PM, and 9% in PM, 5). The literature examples show that CM is usually a mixture of the road (from the abrasion of tyres, linings, brake pads and surface, etc.) and soil dust that is moved by the mixing air masses. Its chemical composition and dust particle sizes change under the influence of various transformations and processes (Sillanpää et al., 2006; Orza et al., 2011). Both the measurement point distance from the traffic sources and considerable seasonal variations in the CM concentrations indicate that the intensity of the soil dust resuspension influenced the CM percentage in the fine dust in Racibórz. The resuspension influence was particularly visible in the spring and summer months (Rogula-Kozłowska et al., 2011). The higher CM percentage in the dust mass observed in the warm season is also related to the fact that the dust concentration was lower then, even though the CM concentration in the air was the same in both seasons.

The average percentage of other elements (OE) in the PM<sub>1</sub> and PM<sub>2.5</sub> masses at the quasi-rural site in Racibórz did not exceed 1% in the cold and warm seasons (Fig. 3). The OE mass was mainly concentrated in PM<sub>1</sub>. The OE mass percentage in PM<sub>1</sub> and PM<sub>2.5</sub> was usually the same. April was an exception as the OE percentage in PM<sub>2.5</sub> was twice as high as in PM<sub>1</sub>.

Elements such as V, Cr, Mn, Co, Ni, Cu, Zn, As, Se, Rb, Sr, Mo, Cd, Sb, Ba, Pb usually make a small percentage of the PM mass (e.g. Querol et al., 2007; Rajšić et al., 2008; Rogula-Kozłowska et al., 2013; Table 2). Nevertheless, they were also important for the analysis because of their great environmental importance due to their toxicity and anthropogenic origin (Almeida et al., 2006). These elements are usually associated with various sources, namely coal combustion, incineration and traffic (Sternbeck et al., 2002; Querol et al., 2007; Ragosta et al., 2008; Rajšić et al., 2008; Rogula-Kozłowska et al. 2013). In Racibórz, the element group is probably related to the local municipal sources. Nonetheless, it is mainly composed of the elements emitted together with other pollutants from distant sources (such as those located in the

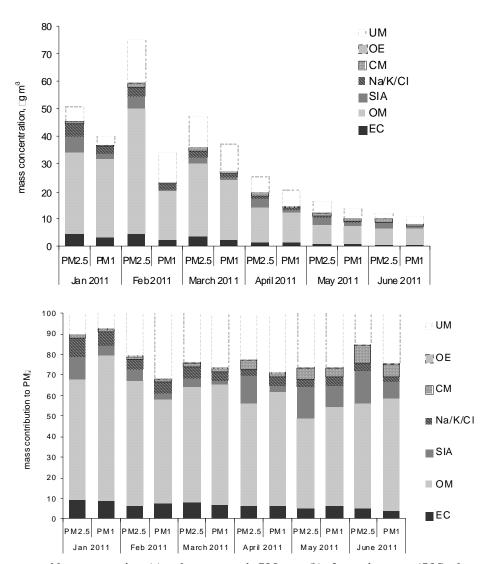


Fig. 3. Average monthly concentrations (a) and percentage in PM mass (b) of organic matter (OM), elemental carbon (EC), secondary inorganic aerosol (SIA), crustal matter (CM), Na/K/Cl, other elements (OE), and of the rest—unidentified matter (UM) related to PM<sub>1</sub> or PM<sub>25</sub> in Racibórz

Upper Silesia urban area). They can move with the air masses into the Racibórz region. Small variations in the OE mass percentage observed in the fine dust mass in the area support this fact.

In the quasi-rural area in Racibórz, the carbonaceous matter (EC+OM) percentage in the PM<sub>1</sub> and PM<sub>2.5</sub> samples was significantly higher than in other European regions (Table 2). On the other hand, the SIA content was visibly lower in the PM samples from Racibórz. The percentages of CM and Na/K/Cl compounds were lower or comparable to those obtained in the locations listed in Table 2. It is worth mentioning that the Na/K/Cl group created for this study was not usually included in the findings from

other locations. Instead, the percentage of sea or road salt in the dust was estimated. The specific emission situation in southern Poland requires separating and discussing the fine dust components directly related to coal and biomass combustion. The results listed in Table 2 suggest that PM concentrations and compositions in Racibórz were mainly influenced by anthropogenic emissions.

### CONCLUSION

The fine dust concentrations can be high in a quasi-rural area in southern Poland. For Racibórz, their mean values were 26.2  $\mu g \cdot m^3$  (PM<sub>1</sub>) and 37.9  $\mu g \cdot m^3$  (PM<sub>25</sub>) in the first half of 2011. They were significantly

higher than those observed in similar European areas. Both PM<sub>1</sub> and PM<sub>2.5</sub> concentrations demonstrated high seasonal variations. The average monthly concentrations in the cold season were at least twice as high as those observed in the warm season. The maximum 24-h concentrations were 124.1 μg·m<sup>-3</sup> (PM<sub>1</sub>) and 152.3 μg·m<sup>-3</sup> (PM<sub>2.5</sub>) in the cold season. The average percentage of PM<sub>1</sub> in PM<sub>2.5</sub> was 69%. The maximum value was found for June (93%). In Racibórz, the OM percentage in the dust mass was 44-61% (PM<sub>2.5</sub>) and 48-71% (PM<sub>1</sub>). It was significantly higher for both fractions in the cold season (increased emissions from fuel combustion). The relatively high OM percentage in the PM mass observed in the warm season mainly resulted from the intensity of the SOC formation in the air, which was higher in the warm season. The highest SIA contents in the PM mass were also observed in the warm season (particularly in May and June). However, the elemental carbon percentage in the dust mass was higher in the cold season. The average EC percentage, observed over the whole measurement period, was ~7% for both PM<sub>1</sub> and PM<sub>25</sub> masses. Additionally, the sum of chloride, potassium and sodium (Na/K/Cl) masses made a significant part of the fine dust mass. It was 3.5-9% for PM<sub>2.5</sub> and 2.4-7%for PM<sub>1</sub>. The CM percentage in the PM<sub>1</sub> and PM<sub>2.5</sub> masses did not exceed 2% in the cold season. The highest CM values were observed in June (9% for PM<sub>2.5</sub> and 6% for PM<sub>1</sub>). The average OE percentage did not exceed 1% in the PM<sub>1</sub> and PM<sub>2</sub> masses in the cold and warm seasons. Even though places such as Racibórz can be formally qualified as rural sites, such areas in southern Poland do not necessarily have low PM concentrations (lower than in cities) or chemical compositions of fine particles different from those in more urbanized areas. The obtained results demonstrate that the fine dust concentrations and compositions in Racibórz in the cold seasons are under the enormous influence of anthropogenic emissions (carbon, waste and biomass combustion in home furnaces and energy production based on hard and brown coal combustion). On the other hand, the researched area is under the influence of the emission from the entire region (including the polluted Upper Silesia and Ostrava areas) in the warm seasons. At that time, the SOC constitutes an important part of the PM mass. The SIA percentage is also significant, but to a lesser extent. The findings indicate serious problems related to the possible reduction in the fine dust concentrations in Racibórz and other areas whose characteristics are similar in the entire southern Poland.

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